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solid iron(ii) chloride is decomposed to produce solid iron and chlorine gas according to the balanced chemical equation: $\text{FeCl}_2(\text{s}) \rightarrow \text{Fe}(\text{s}) + \text{Cl}_2(\text{g})$ what ...

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at constant temperature, 2 l of a gas at 4 em of pressure is expanded to 6 l. what is the new pressure? (do this one conceptually and not algebraically.) d